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Chapter 6. Chemical Bonding. Chemical Bond A link between atoms that results from the mutual attraction of their nuclei for electrons Electrostatic attraction between proton and electron Classified by the way the valence e- are distributed around nuclei of combined atoms Types of Bonds Ionic A chemical bond resulting from electrostatic attraction between cations and anions in a pure ionic bond, one atom completely gives up one or more e- to another atom (rarely happens) Illustration: A + ...

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Modern Chemistry Chapter 6. Chemical Bonding. Chemical Bond. •A link between atoms that results from the mutual attraction of their nuclei for electrons. -Electrostatic attraction between proton and electron. -Classified by the way the valence e- are distributed around nuclei of combined atoms. Types of Bonds. •Ionic.

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Chapter 6 - Chemical Composition

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Modern Chemistry Chapter 6 2 Review Answers

Chapter: Arrangement of Electrons in Atoms. PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. ____ 1. The principal quantum number of an electron is 4.

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